## Instructions for grading the final course project in CHEM 116

The students are working in pairs which are numbered teams (with a few exceptions, Teams \#85, 86, and 87 are individuals who I do not think will do the project because I learned they are withdrawing from the course). When you click on "Needs Grading" in the left navigation, and forward to the end, you will see team numbers and their project submissions. There is one project report for both students on the team. Both students should receive the same score.

When you open the project report to be graded, you will find a screen with assignment details on the right side. To get the rubric option to show up, you have to click the down arrow (see pic at right).

| Assignment Details |  |  |  |
| :---: | :---: | :---: | :---: |
| GRADE |  |  |  |
| LAST GRADED ATTEMPT |  | Group Members $\checkmark$ |  |
| ATTEMPT <br> 12/17/20 10:07 PM <br> To get the rubric to show up, you have to click the down arrow here |  |  |  |
|  |  |  |  |
| SafeAssign $\sim$ |  | 13\% overall match |  |
| SUBMISSION |  |  |  |


| ATTEMPT <br> 12/16/20 8:58 AM |  |
| :--- | ---: |
| SafeAssign $\smile$ | $7 \%$ overall match |

## GRADE BY RUBRIC

FINAL COURSE PROJECT Used for Grading
FEEDBACK TO LEARNER

Item 1

Item 2
Add some overall feedback here

## Grading by rubric

When you open the rubric, there will be places to enter information. I only made rubric descriptions for 0,5 , and 10 out of 10 points for each category. Very likely, you will be giving points that are in between 5 and 10 points. To do this, select the middle item and then in the comment, write how many points out of 10 you gave (you will override the rubric at the bottom). Here is an example of what that looks like:

|  |  | Help ${ }^{5}$ |  |
| :---: | :---: | :---: | :---: |
| Name: Final course project |  |  |  |
| Grid View Lis |  |  |  |
|  | Novice | Competent | Proficient |
| Experimental work | Points: 0 (0.00\%) <br> No photos, no evidence that the expected experimental work was done by both individuals | Points: 5 ( $10.00 \%$ ) <br> Evidence that some of the experimental work was done, or evidence limited to showing that all of it was done by one person and no experimental work was done by the other person | Points: 10 (20.00\%) <br> Photos demonstrating that all aspects of the expected experimental work were carried out and that both individuals participated <br> Feedback: <br> 10 out of 10 . Nice experimental work by both of you. |
| Mathematical work | O Points: 0 ( $0.00 \%$ ) <br> No attempt to do mathematical work | Points: 5 (10.00\%) <br> Mathematical work attempted but parts are incorrect <br> Feedback: <br> 8 out of 10. You used the Nernst equation to show 1.5 V , but didn't say where you got the reduction potentials from. | Points: 10 (20.00\%) <br> Correct usage of the mathematical models related to the project |
| Conceptual work | Points: 0 ( $0.00 \%$ ) <br> No explanations given for the chemistry that the project instructions specify should be explained | Points: 5 ( $10.00 \%$ ) <br> Attempt is made to provide explanations that have relevance, but explanations are unclear, inaccurate, or use chemistry vocabulary that does not make sense <br> Feedback: <br> 8 out of 10. You labeled only half of the items (you were supposed completely map both diagrams to each other). | Points: 10 (20.00\%) <br> Accurate, clear, and logical explanations of chemical ideas (either acid-base or redox) with correct usage of chemical terminology (e.g., conjugate acid/base, pH , oxidation numbers, anode \& cathode) |
| Application work | Points: 0 (0.00\%) <br> Report is absent any mechanisms and/or causal explanations based on kinetic or thermodynamic reasoning | Points: 5 (10.00\%) <br> Kinetics or thermodynamics are invoked but in part are misapplied in reasoning about how to control and/or optimize the chemical system <br> Feedback: <br> 9 out of 10. Increasing $T$ will make the forward and reverse rates faster so battery gets to equilibrium (dead) sooner. Also deltaH is negative (exothermic). But $T$ behavior depends on deltaS sign not deltaH. | Points: 10 (20.00\%) <br> Appropriate application of chemistry models to explain how (mechanism) and/or why (causal explanation) the behavior or function of a chemical system can be controlled and/or optimized |

This is a final project where students are using chemistry they have learned. Pay attention to what they are demonstrating that they learned more than the chemistry they got wrong. Celebrate what students did well, and only take off minimal points for what they did incorrectly (refer to the student handout for specific expectations). In most cases, there will be many things that are incorrect. Pick the one most basic idea that is incorrect and comment on that, so that there can be something learned by the students, and then mostly ignore the rest of what's incorrect. The students are doing real experimental science and, just like you do in your research, it takes a lot of work to figure out if something makes sense. There will also be some things that you won't know how to figure out if they are correct or incorrect. These students are taking a first-year undergraduate course in chemistry, they not in graduate school and they had a very limited time to do these experiments, and they did them at home in their kitchens, not in a lab with equipment, so just ignore things where you have no idea what the students are talking about. Most of what you should be doing is to look for things that they did correctly.

At the bottom of the rubric, leave an overall comment on the rubric. It is fine if you use something similar to what I wrote in the example below, or your own version of it, for most of the students.

Be sure to override the total score (change the number of points) so that it reflects the actual total that you are giving the students. I expect that the vast majority of projects will receive a score somewhere between 40 and 50 .

Raw Total: $\mathbf{3 5 . 0 0}$ (of 50 )
Change the number of points out of 50 to: 45

## Feedback to Learner

For the toolbar, press ALT+F10 (PC) or ALT+FN+F10 (Mac).


You both did an excellent job. It is clear that you worked together and you learned some things. Real experimental work where answers are not known (yet) even to chemists is challenging, and you ran into challenges that you were not able to figure out. That is ok! You were doing real science and you persevered to figure out some things that you did not know before you did this work. The goal was for you to use the chemistry you have learned this semester, and you did that to a great extent.
Path: p Words:93

Do check the SafeAssign score. If you see something that looks very high, please alert me. Also alert me if you read a report where it is clearly indicated that only one student in the pair did the work. I will grade these situations.

The score you give in the rubric will automatically populate to both students in the team. Do not fill in the "Group members" section at the end because that overrides the score you gave. On the next page is what the screen looked like when I returned after saving the rubric and before I submitted.

In the section on "Feedback to learner", I directed students to look at the rubric, since they will need to click on the rubric in Blackboard in order to see their scores and the feedback, and they might not know to do that.

Afterward, click "Submit" and the grade should then go to the students.


